

Multiple Choice Questions

1	B
2	C
3	C
4	C
5	B
6	D
7	A
8	B
9	A
10	A
11	A
12	A
13	D
14	C
15	C

- 1 a i Bung at the top of the fractionating column
ii fractionating column
iii separates the components
iv turns vapour to liquid
- b i 141
ii propanoic acid
iii temperature rises
- 2 a i C
ii D
iii B
iv A
v D
- b i 20
ii consists of two atoms only

- iii atoms of the same elements having the same number of protons but different number of neutrons
- c 17
 ^{37}Cl
20
35
46
 $^{85}\text{Rb}^+$
- 3 a M – hydrochloric acid
N – Manganese dioxide
R – sulfuric acid
- b 3
- c upward delivery
- 4 a number of shells corresponds to period number and number of outermost electrons correspond to group number
- b 2,8
- c i sharing of 2 pairs of electrons
ii Since Mr of N_2 (28) is smaller than Mr of O_2 (32), nitrogen will diffuse faster than oxygen
- 5 a Two Na atoms lose 1 electron each. These electrons are transferred to 1 oxygen atoms.
- b In molten Na_2O , the ions are mobile whereas in solid state ions can only vibrate about fixed positions.
- c The ionic bonds in MgO are stronger than in Na_2O due to higher charge in Mg^{2+}
- d i spreading of particles from a region of high concentration to a region of low concentration
ii at higher temperature, particles gain kinetic energy and move faster

iii They have different molecular masses

- 6 a Chromatography
- b To prevent the colourings from dissolving in the solvent
- c The pen also contains coloured inks
- d S
- e R and T
- f $R_f = \frac{\text{distance travelled by spot}}{\text{distance travelled by solvent}}$
- g The value must be less than 1
- h All spots are coloured